

Diffusion of organic molecules on metals

System	E_m [meV]	E_d [meV]	D_0 [cm ² /s]	θ	T [K]	Method	Ref.	E_b [eV]	Ref.
C₃H₈/Ru(0001)		130 ± 10	0.11	0.2	95-115	LITD	[1]	0.48	[1]
n-C₄H₁₀/Ru(0001)		150 ± 10	0.11	0.2	110-140	LITD	[1]	0.52	[1]
n-C₅H₁₂/Ru(0001)		200 ± 10	0.30	0.2	130-160	LITD	[1, 2]	0.60	[1]
n-C₆H₁₄/Ru(0001)		210 ± 10	0.16	0.2	140-190	LITD	[1]	0.65	[1]
isopentane/Ru(0001)		180 ± 10	$5.5 \times 10^{-2 \pm 0.1}$	0.1	130-160	LITD	[2]	0.59	[2]
cyclo-pentane/Ru(0001)		140 ± 10	$6.0 \times 10^{-4 \pm 0.1}$	0.1	130-160	LITD	[2]	0.52	[2]
neopentane/Ru(0001)		130 ± 10	$4.0 \times 10^{-2 \pm 0.1}$	0.1	130-160	LITD	[2]	0.46	[2]
perfluoro-n-pentane/Ru(0001)		130 ± 10	$5.9 \times 10^{-2 \pm 0.2}$	< 0.1	95-120	LITD	[3]	0.56	[3]
tetramethylsilane/Ru(0001)		140 ± 5	$5.9 \times 10^{-2 \pm 0.1}$	0.4	100-130	LITD	[4]	0.53	[4]
C₂H₂/Pd(110)	~ 570		{ 8×10^{-4} }	monomer	250-260	STM	[5]	–	
C₂H₂/Pd(111)	~ 180		{ 2×10^{-3} }	monomer	65-70	STM	[6]	0.36	[6]
C₆H₆/Pd(110)	~ 570 \perp		{ 8×10^{-4} }	monomer	210-230	STM	[7]	1.1±0.2	[8]
PVBA/Pd(110)	830±30		$7.6 \times 10^{-6 \pm 0.4}$	monomer	330-370	STM	[9]	–	
C₆₀/Pd(110)	~ 1500		–	monomer	440-480	STM	[10, 11]	–	
C₂H₂/Cu(001)	530±10		$1.6 \times 10^{-2.4 \pm 0.2}$	monomer	180-210	STM	[12]	–	
CH₄/Pt(111)		~ 23	{ 3.3×10^{-4} }	low	45	LITD	[13]	0.16	[13]

θ is given in terms of the respective saturation coverage; estimates from original authors in {};
|| is the [1-10] and \perp the [001] direction on a fcc(110) substrate.

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